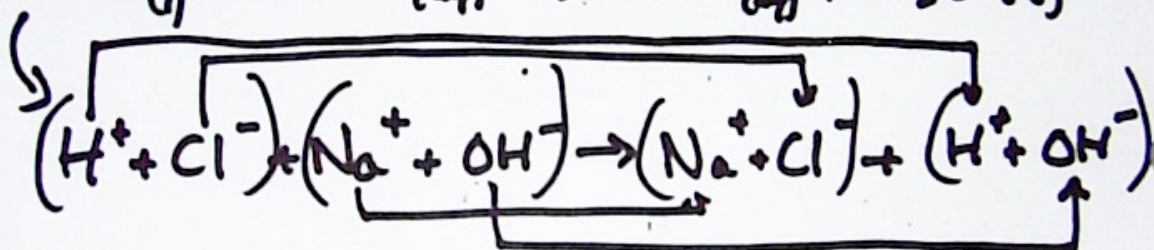
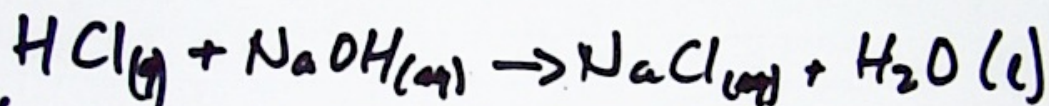


SAMPLE Problem 19.5

* Calculating K_a for a weak acid

* ICE TABLE

Neutralization Reactions



Titration neutral when $[\text{H}^+] = [\text{OH}^-]$

or # moles Acid = # moles Base

